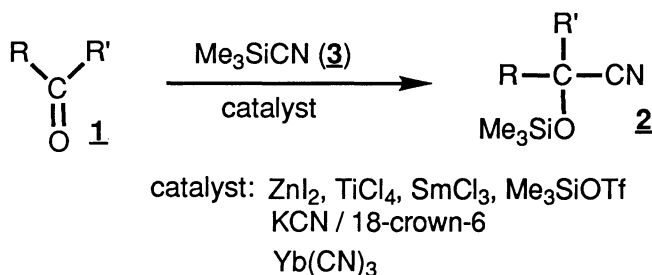


Ytterbium Tricyanide, a Highly Efficient Catalyst for
the Addition of Cyanotrimethylsilane to Carbonyl Compounds

Seijiro MATSUBARA, Tsutomu TAKAI, and Kiitiro UTIMOTO*
Department of Industrial Chemistry, Faculty of Engineering,
Kyoto University, Yoshida, Kyoto 606-01

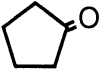
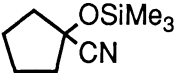
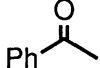
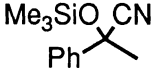
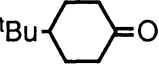
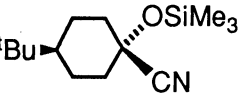
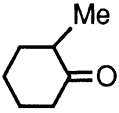
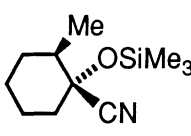
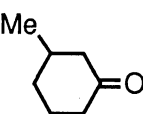
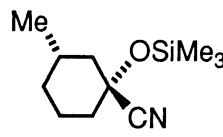
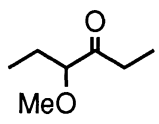
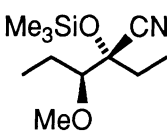
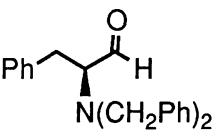
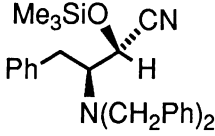
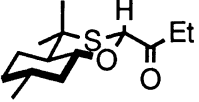
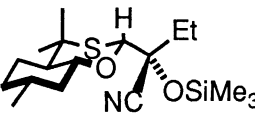
Reaction of cyanotrimethylsilane with various carbonyl compounds was effectively catalyzed by $\text{Yb}(\text{CN})_3$ to give the adducts in excellent yields; reaction with substituted cyclohexanones proceeded in a highly stereoselective manner.

α -Trimethylsiloxy nitriles **2**, trimethylsilyl ether of cyanohydrins, are useful intermediates in organic synthesis and have been prepared by the reaction of a carbonyl compounds **1** with cyanotrimethylsilane (**3**).¹⁻⁸⁾ Various types of catalysts have been employed: Lewis acids, such as ZnI_2 and TiCl_4 , are generally used,^{1,2,4)} and lanthanoid salts have been applied.⁵⁾ Trimethylsilyl triflate also showed a high catalytic activity.⁶⁾ On the other hand, KCN-crown ether system generates highly reactive cyanating species and can be employed to the reaction with easily enolizable ketones.⁷⁻⁹⁾ We have reported that lanthanoid cyanides are highly effective catalysts for the reaction of **3** with oxiranes¹⁰⁾ or aziridines.¹¹⁾ In these reactions, $\text{Yb}(\text{CN})_3$ showed little Lewis acidity, although it worked as powerful cyanating reagent.¹⁰⁾ These observations prompted us to anticipate that $\text{Yb}(\text{CN})_3$ might possess high catalytic activity for the above described cyanohydrin formation. This paper describes the successful application of $\text{Yb}(\text{CN})_3$ as a catalyst for the reaction of **3** with various types of carbonyl compounds **1**.



Reaction of **3** with ketones proceeded smoothly under $\text{Yb}(\text{CN})_3$ catalysis. To a mixture of $\text{Yb}(\text{CN})_3$ (50 mg, 0.2 mmol) and **3** (2.4 mmol) in THF (5 mL), acetophenone (**1b**, 2.0 mmol) was added at room temperature and the reaction mixture was stirred for 1 h. Hydrolytic workup gives the adduct **2b** in 99% yield.¹²⁾ Various ketones reacted similarly with **3** to give the corresponding **2** in excellent yields and in high diastereoselectivity. The combinations of ketone **1**, reaction temperature, reaction time, yield of the product **2**, as well as the

Table 1. Yb(CN)₃ Catalyzed Reaction of Ketones with Me₃SiCN^{a)}

Ketone (1)	Conditions		Product (2)	Yield ^{b)} % (%de)	ZnI ₂ ^{c)}	Me ₃ SiOTf ^{d)}	KCN/ crown ^{e)}
	Time / h	Temp / °C					
a 	1.5	0		93	94	>99	99
b 	12	25		>99	91	88	>99
c 	2	0		88 (88) [>99 (94)] ^{f)}	97 (80)	99 (82)	99 (56)
d 	3.5	0		87 (86) [92 (90)] ^{f)}	94 (32)	>99 (31)	90 (46)
e 	2	0		94 (54) [90 (94)] ^{f)}	94 (2)	>99 (4)	90 (16)
f 	5	0		81 (56)	82 (64)	85 (58)	80 (10) ^{g)}
g 	1	0		91 (44)	89 (82) ^{h)}	99 (82)	92 (72)
h 	1.5	0		>99 (70) [99 (74)] ⁱ⁾	99 (44)	74 (44)	98 (26) ^{g)}

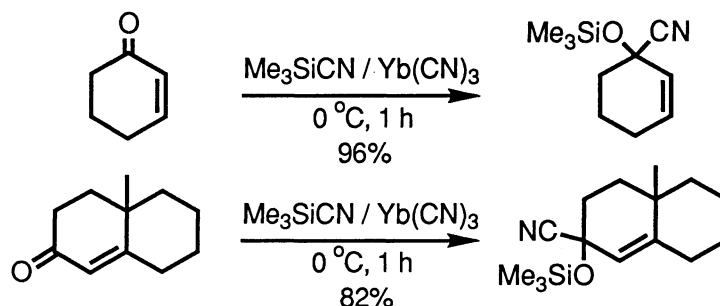
a) To a slurry of Yb(CN)₃ (0.1 mmol) in THF (4 mL), **3** (1.2 mmol) and a ketone (1.0 mmol) was added. The mixture was stirred at the indicated temperature. b) Isolated yields. c) ZnI₂, Ref.1 and 2. d) Me₃SiOTf, Ref.6.

e) KCN/18-crown-6, Ref.8. f) Reaction at -89 °C. g) Major product is the diastereomer of the isomer shown in Table.

h) See Ref.4. i) Reaction at -23 °C.

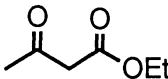
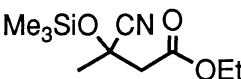
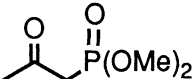
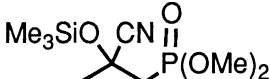
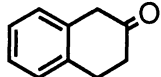
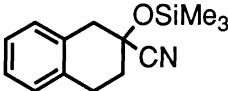
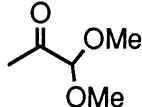
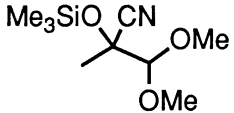
yield of the reported methods¹³⁾ are summarized in Table 1. Reaction of **3** with 4-*t*-butylcyclohexanone (**1c**) completes at -89 °C within 6 min in highly diastereoselective manner to give **2c** in >99% (94% d.e.).

Reaction of α,β -unsaturated ketones with **3** afforded 1,2-adducts selectively at 0 °C; the reaction in refluxing THF gave the same 1,2-adducts.



Reaction with easily enolizable ketones (**1i** - **1k**) or a ketone (**1l**) containing acid-sensitive group afforded the corresponding adducts in good yields (Table 2). As can be seen from Table 2, $\text{Yb}(\text{CN})_3$ catalyst gave the corresponding adducts in good yields as analogous to the KCN-crown ether catalyst system, whereas ZnI_2 could not be employed in these cases.

Table 2. $\text{Yb}(\text{CN})_3$ Catalyzed Reaction of Enolizable Ketones with Me_3SiCN ^{a)}

Ketone (1)	Condition		Product (2)	Yield ^{b)} %	ZnI ₂ ^{c)}	Me ₃ SiOTf ^{d)}	KCN/ crown ^{e)}	
	Time / h	Temp / °C						
i		4	25		90	45 ^{f)}	42 ^{f)}	73
j		5.5	25		84	0	0	96
k		2	0		88	68	94	84
l		1	0		91	0	30	83

a) To a slurry of $\text{Yb}(\text{CN})_3$ (0.1 mmol) in THF (4 mL), **3** (1.2 mmol) and a ketone (1.0 mmol) was added. The mixture was stirred at the indicated temperature. b) Isolated yields. c) ZnI_2 , Ref.1 and 2. d) Me_3SiOTf , Ref.6.

e) KCN/18-crown-6, Ref.8. f) Silyl enol ether was isolated in 50% yield (ZnI_2) and 48% yield (Me_3SiOTf).

The results shown in Table 1 and Table 2 indicates that $\text{Yb}(\text{CN})_3$ can be widely employed as the catalyst for the addition of **3** with various carbonyl compounds.

Some remarks on $\text{Yb}(\text{CN})_3$ are added lastly. Ytterbium tricyanide could be obtained by the following two methods: (1) $\text{Yb}(\text{CN})_3$ was prepared by the reaction of **3** with $n\text{-Bu}_3\text{Yb}$ which was produced from YbCl_3 and 3 equiv. of $n\text{-BuLi}$. IR showed two prominent peaks at 2210 (medium) and 2110 (strong). $\text{Yb}(\text{CN})_3$ thus obtained showed a high catalytic activity in the reaction of carbonyl compounds **1** with **3**. (2) $\text{Yb}(\text{CN})_3$ was also obtained by the reaction of $\text{Yb}(\text{O}^i\text{Pr})_3$ with 3 equiv. of **3**. IR showed a strong peak at 2085. $\text{Yb}(\text{CN})_3$ thus obtained showed a high activity in the reaction of **3** with **1** but much less active in the reaction of **3** with aziridines; the reactivity of $\text{Yb}(\text{CN})_3$ was enhanced by the addition of LiCl or LiCN .^{14,15)} Structure of the individual sample as well as the relationships between the absorptions and reactivities are still open to be answered.¹⁶⁾

References

- 1) P. G. Gassman and J. J. Talley, *Tetrahedron Lett.*, **1978**, 3773.
- 2) D. A. Evans, G. L. Carroll, and L. K. Truesdale, *J. Org. Chem.*, **39**, 914 (1974).
- 3) K. Utimoto, Y. Wakabayashi, T. Horie, M. Inoue, Y. Shishiyama, and H. Nozaki, *Tetrahedron*, **39**, 967 (1983) and references cited therein.
- 4) M. T. Reetz, M. W. Drewes, K. Harms, and W. Reif, *Tetrahedron Lett.*, **29**, 3295 (1988). and references cited therein.
- 5) A. E. Vougiokas and H. B. Kagan, *Tetrahedron Lett.*, **28**, 5513 (1987).
- 6) R. Noyori, S. Murata, and M. Suzuki, *Tetrahedron*, **37**, 3899 (1981).
- 7) R. N. Comber, C. A. Hosmer, and W. J. Brouillette, *J. Org. Chem.*, **50**, 3627 (1985).
- 8) W. J. Greenlee and D. E. Hangauer, *Tetrahedron Lett.*, **24**, 4559 (1983).
- 9) Structure of the reactive species of 18-crown-6/ KCN -**3** is assigned as $\text{Si-N}=\text{C}$ based on a theoretical calculation of IR absorption; M. B. Sassman, G. K. Prakash, and G. Olah, *J. Org. Chem.*, **55**, 2016 (1990).
- 10) S. Matsubara, H. Onishi, and K. Utimoto, *Tetrahedron Lett.*, **31**, 6209 (1990).
- 11) S. Matsubara, T. Kodama, and K. Utimoto, *Tetrahedron Lett.*, **31**, 6379 (1990).
- 12) Reaction of PhCOMe (**1b**) with **3** under SmCl_3 catalysis afforded the corresponding adduct **2b** in 50% yield, under $\text{Eu}(\text{fod})_3$ catalysis gave a mixture of **2b** (25%) and the corresponding protodesilylated nitrile (29%); reaction of PhCHO with **3** afforded the adduct in 98% yield.⁵⁾
- 13) Reactions were carried out under the conditions described in references: ZnI_2 -catalyzed reaction;¹⁾ Me_3SiOTf catalyzed reaction;⁶⁾ reactions catalyzed by $\text{KCN}/18\text{-crown-6}$.^{7,8)}
- 14) S. Matsubara, T. Kodama, and K. Utimoto, in preparation.
- 15) $\text{Yb}(\text{CN})_3$ was prepared by the reaction of YbBr_3 with LiCN ; K. Rossmannith, *Monatsh. Chem.*, **97**, 1698 (1966); IR absorptions, 2185 (weak) and 2100 (medium).
- 16) The authors thank to the Japan Society for the Promotion of Science for a Grant-in-Aid for International Exchange Program.

(Received May 20, 1991)